

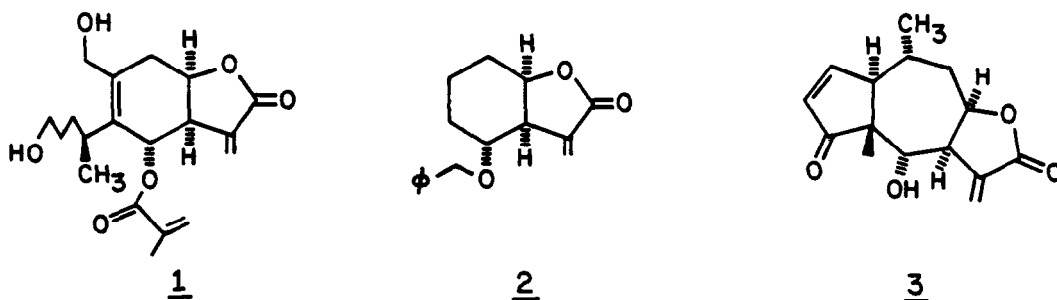
AN APPROACH TO ERIOLANIN. SYNTHESIS OF α -METHYLENE LACTONES
OF TRANS-1,3-DIHYDROXYCYCLOHEXANES

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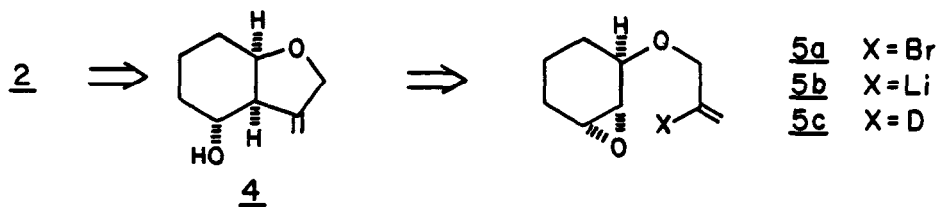
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Abstract: A new synthesis of α -methylene lactones is discussed which incorporates an intramolecular epoxide opening with a vinyl anion.

As part of our proposed synthesis of eriolanin (1),¹ we required methodology for the synthesis of the lactone portion of 1 which incorporated stable, latent functionality that could be easily transformed into the α -methylene lactone moiety. Although numerous methods exist for the synthesis of α -methylene- γ -butyrolactones fused to cyclohexanes,^{2a} a few direct methods are available for the preparation of α -methylene- γ -butyrolactones of trans-1,3-dihydroxycyclohexanes.^{2b} We wish to report the successful synthesis of 2, a model for the lactone portion of 1. The α -methylene lactone of a trans-1,3-diol is a structural unit also found in the pseudoquaianolide helenalin (3),³



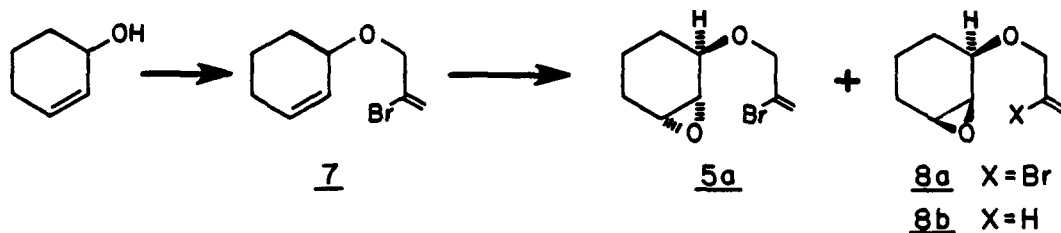
Our plan was to use an intramolecular epoxide opening of vinyl anion 5b to give 4. We felt that allylic ether 4 would serve as stable latent functionality which could be easily converted to the α -methylene lactone 2 by selective allylic oxidation.



The initial attempts to synthesize 5a were not promising. Alkylation of epoxy alcohol 6⁴ using 2,3-dibromopropene under a variety of different conditions⁵ gave only low yields (5-10%) of vinyl bromide 5a. However, using



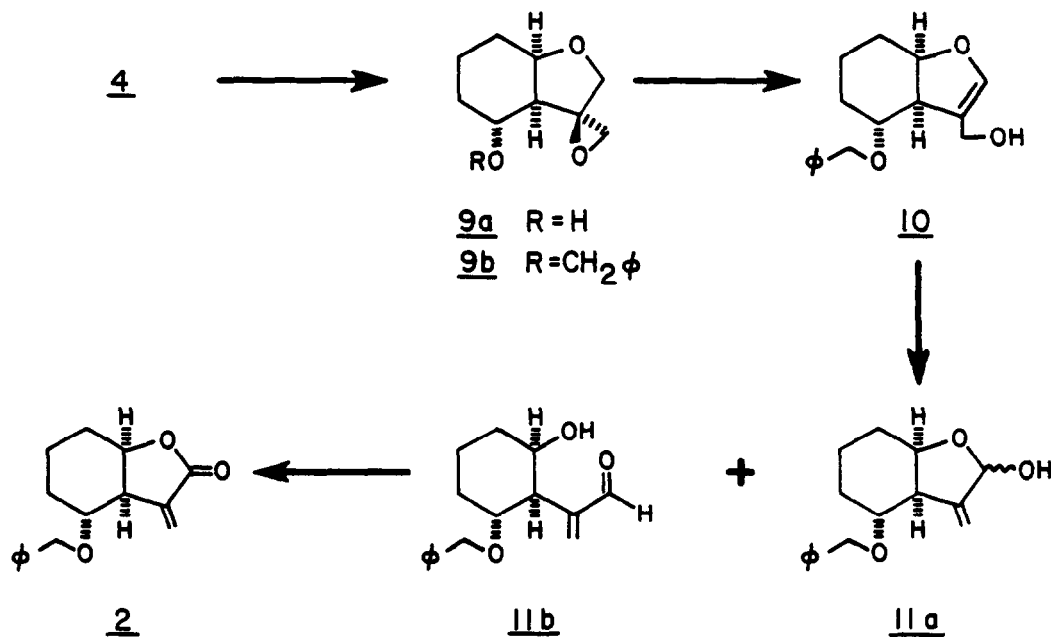
solvolytic conditions, epoxy bromide 5a could be prepared in good yield. Treatment of 2-bromo-2-propenol⁶ with cyclohexenol (2.0 equiv. in 3 portions) and a catalytic amount of *p*-toluenesulfonic acid in benzene at 65°C for 18 h gave an 81% yield of ether 7. Epoxidation (mCPBA, 25°C, CCl₄, 12 h)



gave rise to an 85% yield of a 7:1 mixture of desired epoxide 5a and the corresponding *cis* isomer 8a.

The cyclization/epoxide opening proved troublesome.⁷ The conversion of 5a to 4 was first attempted through the intermediacy of lithium reagent 5b. Initial experiments led only to the formation of allyl ether 5c after quenching the anion (tBuLi, Et₂O, -78°C) with a solution of D₂O in THF. None of the desired bicyclic alcohol was obtained even after 8 h at -78°C. Epoxy alcohol 6 was the only product formed at higher temperatures (-40°C, -20°C). However, we found that the cuprate formed from vinyl bromide 5a produced the desired result.⁸ Treatment of vinyl bromide 5a with 4 equiv of nBu₂CuLi (2 h, Et₂O) at -78°C provided alcohol 4 in 89% yield. (Under the same conditions, cis epoxy bromide 8a gave rise to ether 8b.) ¹H NMR (270 MHz) and ¹H NOE difference experiments confirmed the stereochemistry of 4.⁹ The successful cyclization/epoxide opening of 5a to give alcohol 4 establishes the cis ring fusion and the trans-1,3-diol functionality necessary for the synthesis of 1.

Conversion of 4 to α-methylene lactone 2 was carried out in the following way. Directed epoxidation (VO(acac)₂, tBuOOH, 25°C, 3 h)¹⁰ in chloroform gave epoxide 9a (80% yield) as a single isomer, which was subsequently converted to benzyl ether 9b (THF, NaH, HMPA (4x), φCH₂Br, 16 h, 25°C, 91%). Base catalyzed



rearrangement of the epoxide¹¹ using lithium diethylamide (3 equiv, Et₂O, 0°C, 4 h) yielded allylic alcohol 10 (61% yield).^{12,13} Acid catalyzed rearrangement (5% H₂SO₄, THF, 30 min, 25°C) of 10 gave a mixture of 11a and 11b (90% yield) which was oxidized (MnO₂, CH₂Cl₂, 15h, 25°C)¹⁴ to provide α-methylene lactone 2 (74% overall yield from 10).¹⁵

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 - ¹H NMR data **4** (CDCl₃): δ5.12 (bs, 1H); 4.94 (bs, 1H); 4.5 (d, J = 14 Hz, 1H), 4.31 (dt, J = 14, 2.5, 1H), 4.05 (dd, J = 3.5, 7.0, 1H) 3.46 (ddd, J = 10, 10, 4.5, 1H), 2.27 (dd, J = 10, 4.5, 1H), 1.9 (m, 2H), 1.6 (m, 3H), 1.35 (m, 1H).
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 - Base catalyzed rearrangement of epoxide **i** [Prepared in 91% overall yield (5:1 mixture of **i**:**9b**) by the alkylation of **4** (NaH, THF, φCH₂Br, HMPA, 25°C, 8 h) followed by epoxidation (mCPBA, CCl₄, 25°C, 6 h)] gave a 2:1 mixture of allylic alcohols **10** and **ii**.
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- Ca. 10% of amine **iii** was observed in the ¹H NMR of the crude reaction mixture.
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